Benzenemethanethiol CAS Number 100-53-8

201-15166A

High Production Volume (HPV) Challenge Program

04 APR - 9 PM 12: 5:

Benzenemethanethiol CAS Number 100-53-8

Chevron Phillips Chemical Company LP

10001 Six Pines Drive The Woodlands, Texas 77380

March 2004

TABLE OF CONTENTS

I.	Executive Summary4					
II.	General Substance Information6					
III.	Structural Surrogate Discussion					
IV.						
V.	Evaluation of Environmental Fate Data					
	A. Photodegradation – Atmospheric Oxidation	9				
	B. Hydrolysis					
	C. Chemical Transport and Distribution in the Env					
	(Fugacity Modeling)					
	D. Biodegradation and Bioaccumulation					
VI.	Ecotoxicity Data	11				
VII.	Mammalian Toxicity	12				
	A. Acute Toxicity					
	B. Repeated Dose Toxicity					
	C. Genetic Toxicity/Mutagenicity					
	D. Reproductive Toxicity					
	E. Developmental Toxicity					
VIII.	I. "Beyond Sids" Endpoints1					
IX.	Conclusions15					
v	Deferences 17					

LIST OF TABLES AND FIGURES

Table 1: Matrix of Available and Adequate Data on BZM and PHM

Measured (M) and Calculated (C) Physicochemical Properties Table 2:

Results for Environmental Fate and Pathways Table 3: Table 3a: EPIWIN Level III Fugacity Results for BZM

Table 3b: EPIWIN Level III Fugacity Results for PHM

Table 4: Results for Ecotoxicity Endpoints

Table 5: Results for Mammalian Toxicity Endpoints

Figure 1: Structural Comparison of Benzenemethanethiol and Benzenethiol

LIST OF APPENDICES

Appendix I: International Uniform Chemical Information Dataset for BZM Appendix II: International Uniform Chemical Information Dataset for PHM

ABBREVIATIONS

BCF = predicted bioconcentration factor

Benzenemethanethiol = Benzyl Mercaptan

BZM = Benzyl Mercaptan

 $cm^3 = cubic centimeter$

CPChem = Chevron Phillips Chemical Company LP

CSI-Closed System Intermediate

gd = gestation day

 H_2S = hydrogen sulfide

hPa = hectopascal

HPV = High Production Volume

IUCLID = International Uniform Chemical Information Dataset

Koc = organic carbon partition coefficients

 LC_{50} = lethal concentration (to 50% of animals dosed)

 LD_{50} = lethal dose (to 50% of animals dosed)

LOAEL = lowest observed adverse effect level

mg/kg = milligram per kilogram

mg/L = milligrams per liter

mmHg = millimeter mercury

NaSH = sodium hydrosulfide

NOAEL = no observed adverse effect level

OECD = Organisation for Economic Cooperation and Development

PHM = Phenyl Mercaptan

QSAR = Quantitative Structure Activity Relationship

RACB = Reproductive Assessment by Continuous Breeding

SIDS = Screening Information Data Set

USEPA = United States Environmental Protection Agency

USFDA = United States Food and Drug Administration

I. EXECUTIVE SUMMARY

Chevron Phillips Chemical Company LP (CPChem) is committed to fulfilling the High Production Volume (HPV) commitments it made under the United States Environmental Protection Agency (USEPA) HPV Challenge Program on February 14, 2001. As part of this commitment, CPChem has volunteered to assess the health and environmental hazards, including selected physicochemical characteristics of Benzenemethanethiol (CASN 100-53-8), commonly known as and referred to hereafter as Benzyl Mercaptan (BZM). BZM is a member of the broader family of organomercaptans, which are characterized by their very strong malodor and low odor threshold. CPChem currently manufactures BZM for use as a raw material in the production of agricultural pesticide ingredients.

CPChem has identified data from company proprietary files, peer-reviewed literature, and/or calculated endpoints using widely accepted computer modeling programs. In fulfillment with USEPA guidance for use of read-across data (USEPA, 1999b), CPChem proposes the use of surrogate data from Phenyl Mercaptan (CASN 108-98-5) (PHM), a close structural analogue to BZM, to fill Screening Information Data Set (SIDS) endpoint data gaps and provide additional support in our understanding of health and environmental hazards for BZM. Both BZM and PHM have a relatively narrow range of physicochemical parameters and are composed of similar functional groups. Thus, these two substances are expected to demonstrate similar environmental fate and toxicological profiles.

Physicochemical endpoints for BZM are fulfilled by using existing measured data or data calculated by the EPIWIN[©] computer model. No additional testing is proposed for this program.

An estimation from a Level III fugacity model predicts that BZM and PHM will likely partition to soil and water. Ready biodegradation testing showed that BZM is not readily biodegradable. The predicted bioconcentration factors and organic carbon partition coefficients for BZM and PHM suggest similar fate profiles in the environment and no bioaccumulation hazard for either BZM or PHM. A review of the existing data for BZM and PHM shows that sufficient data are available to characterize the environmental fate of BZM, with the exception of the Hydrolysis SIDS endpoint. Additional testing for the Hydrolysis SIDS endpoint is therefore proposed for this program.

A review of the existing data for BZM and PHM shows that insufficient data are available to characterize aquatic toxicity. Additional testing for the Acute Toxicity to Fish and Aquatic Plants (Algae) endpoint is proposed for this program.

A limited amount of existing mammalian toxicity information on BZM and PHM demonstrates that PHM is a conservative read across benchmark, with slightly higher toxicity than BZM. Acute toxicity studies show that BZM is of low acute toxicity by oral, inhalation, dermal, and intraperitoneal routes. BZM has not been tested for reproductive toxicity; reproductive toxicity data are available for PHM. Due to its close

structural similarity to PHM, as well as the demonstrated higher level of acute toxicity of PHM, it would be expected that BZM would be of a similar order of magnitude as PHM, if not of lower reproductive toxicity. Repeated dose toxicity testing has not been performed on BZM or PHM; therefore, repeated dose toxicity testing is proposed. Likewise, no *in vitro* chromosomal aberration studies were identified for either BZM or PHM. As a result, testing is proposed to meet the chromosomal aberration requirements. BZM has not been tested for developmental toxicity; however data is available for PHM. Due to its close structural similarity to PHM, as well as the demonstrated higher level of acute toxicity of PHM, it would be expected that BZM would be of a similar order of magnitude as PHM, if not of lower developmental toxicity.

Nearly all HPV endpoints have been satisfied for BZM. The close structural surrogate PHM serves as a read-across for some endpoints (as described below). PHM surrogate data are available for several HPV endpoints. Table 1 summarizes the available data for BZM and PHM.

Table 1. Matrix of Available and Adequate Data on BZM and PHM

Test	BZM Y/N (Klimish Score)	PHM Y/N (Klimish Score)	Testing Planned? Y/N
Physical and Chemical Data			
Melting Point	Y (2)	Y (2)	N
Boiling Point	Y (2)	Y (2)	N
Vapor Pressure	Y (2)	Y (2)	N
Partition Coefficient	Y (2)	Y (2)	N
Water Solubility	Y (2)	Y (2)	N
Environmental Fate and Pathways			
Photodegradation	Y (2)	Y (2)	N
Stability in Water (Hydrolysis)	N	N	Υ
Transport/Distribution	Y (2)	Y (2)	N
Biodegradation	Y (2)	N	N
Ecotoxicity			
Acute/Prolonged Toxicity to Fish	N	N	Υ
Acute Toxicity to Aquatic Invertebrates (Daphnia)	Y (1)	N	N
Acute Toxicity to Aquatic Plants (Algae)	N	N	Υ
Toxicity			
Acute Toxicity (Oral)	Y (2)	Y (2)	N
Acute Toxicity (Inhalation)	Y (2)	Y (2)	N
Acute Toxicity (Dermal)	Y (1)	Y (2)	N
Repeated Dose	N	N	Υ
GeneticToxicity in vitro –	Y (2)	N	N
Gene Mutation			
Genetic Toxicity –			
Chromosomal Aberration	N	N	Υ
Reproductive Toxicity	N	Y (1)	N
Developmental Toxicity	N	Y (1)	N

Note: The data used to characterize the OECD SIDS endpoints for substances in this Test Plan were identified either in company proprietary files, peer-reviewed literature, and/or calculated using widely accepted computer modeling programs. PHM was used for read-across as defined by the USEPA (1999b). All data were evaluated for study reliability in accordance with criteria outlined by the USEPA (1999a). Only studies that met the reliability criteria of "1" (reliable without restrictions) or "2" (reliable with restrictions) were used to fulfill OECD SIDS endpoints. Additional data for BZM and PHM are also included in the IUCLID (International Uniform Chemical Information Dataset) attached in Appendices I and II. A more detailed discussion of the data quality and reliability assessment process used to develop this test plan is provided in Appendix III.

Summary: Adequate data (i.e., Klimisch rating 1 and 2) are available for all endpoints except Hydrolysis, Acute Toxicity to Fish, Acute Toxicity to Aquatic Plants, Repeated Dose, and Genetic Toxicity: *In vitro* Mammalian Cytogenetic Test. Additional testing with BZM is proposed to fulfill these six endpoints.

II. GENERAL SUBSTANCE INFORMATION

BZM is a member of the broader family of organomercaptans, which are also sometimes referred to as thiols or sulfhydryls. Many are naturally occurring and are characterized by their very strong malodor and low odor threshold (~1 ppb), a property that creates an immediate odor nuisance while keeping exposure and potential for adverse effects to humans at a minimum.

Today, CPChem is the major US producer of BZM, and CPChem annual production volumes are sold to one client. BZM is manufactured and transported in a closed system process and used as a chemical intermediate to produce agricultural pesticides, also in a closed system process. Traces or residues of BZM in end products are negligible as ppb amounts would introduce a malodor.

CPChem originally prepared documentation to substantiate that BZM is a closed system intermediate (CSI) pursuant to the USEPA's guidance for closed system intermediates for the HPV Challenge Program. However, CPChem dropped this approach based on the following two factors:

- It is unlikely the USEPA and the international community will accept closed system intermediate status and a reduced test package for BZM because CPChem's processing of BZM in the US is not site limited. The BZM is transported by railcar to another site, where it is a chemical intermediate in the production of agricultural pesticides.
- BZM is also approved by U.S. Food and Drug Administration (FDA) as a food additive. This is suggestive that specialty chemical distributors marketing natural BZM for this application may exist.

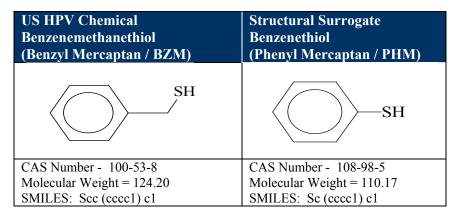
BZM CSI processes are confidential business information, but are available upon request for further evaluation by USEPA.

III. STRUCTURAL SURROGATE DISCUSSION

Because a substantial amount of data exists for BZM, it is possible to characterize several of the OECD SIDS hazard endpoints and to meet the requirements of the USEPA HPV Challenge Program. However, PHM also serves as a close structural surrogate to BZM and adds to the weight of evidence in characterizing many BZM physical/chemical, environmental, and human health-related endpoints.

Figure 1 presents the chemical structures for BZM and PHM. These substances are close structural analogs having the same functional groups. They differ only in the presence in BZM of a methylene group between the aromatic ring and the thiol group, which results in a difference in molecular weight of only 14 atomic mass units. The following sections illustrate that these surrogates have similar physical, chemical, environmental fate and effects, and toxicological properties. Given that PHM is slightly lower in molecular weight, it is slightly more water soluble, mobile, and bioavailable, making it a conservative read across benchmark with slightly higher toxicity than BZM. This is important because the trend is consistent and shows that where there are BZM data gaps, and PHM data are available for read-across purposes, additional testing on BZM will not provide new useful data.

Figure 1. Structural Comparison of Benzenemethanethiol and Benzenethiol



IV. PHYSICOCHEMICAL PROPERTIES

The physical chemical data for BZM and PHM provided in Table 2 were experimentally confirmed or primarily obtained from well-established and scientifically accepted reference handbooks such as the Merck Index (O'Neil, 2001), Patty's Industrial Hygiene and Toxicology (Bingham, 2001), Sax's Dangerous Properties of Industrial Materials (Lewis, 2000), and the CRC Handbook of Chemistry and Physics (Lide, 2001-2002), as

well as EPIWIN-calculated values (USEPA and Syracuse Research Corporation, 2000). These data show that both BZM and PHM are moderately water soluble and have similar melting and boiling points, vapor pressure, and hydrophobicity (log Kow).

Table 2. Measured (M) and Calculated (C) Physicochemical Properties

Physical and Chemical Data					
Test	M/C	BZM	M/C	PHM	
Melting Point			M ^{3,4}	-14.8 to -14.9 °C	
	M^1	-30 °C	M^1	-14.9 °C	
	C ²	-19.22 °C	C ²	-31.86 °C	
Boiling Point	$M^{4,5}$	194-195 °C	$M^{3,4,6}$	168.7 to 169.5 °C	
	M_{\cdot}^{1}	194.5 °C	M_{\cdot}^{1}	169.1 °C	
	C ²	200.14 °C	C^2	176.14 °C	
Vapor			M ⁴	2.67 hPa at 25 °C	
Pressure			M_{\perp}^{1}	1.93 mmHg (2.57 hPa)	
	C ²	0.474 mmHg (0.632 hPa)	C ²	1.63 mmHg (2.17316 hPa)	
Kow Partition			M ⁸	2.52	
Coefficient			M ¹	2.52	
Coomoran	C ⁷	2.48	C ⁷	2.69	
Water			M^3	1 at 25 °C	
Solubility			M^1	835 mg/L (at 25 °C)	
-	C ₉	732.2 mg/L (at 25 °C)	C ₉	765.5 mg/L (at 25 °C)	

¹EPIWIN v3.10; measured values from the EPIWIN experimental database.

Summary: Adequate data (i.e., Klimish rating 1 and 2) are available for all endpoints; no additional testing is proposed for the USEPA HPV Challenge Program (see Table 2 and IUCLID documents).

V. EVALUATION OF ENVIRONMENTAL FATE DATA

Environmental fate data for BZM and PHM were either experimentally measured or estimated using EPIWIN, and are provided in Tables 3, 3a, and 3b. Overall, these substances are expected to be mobile if released to the environment, but they will ultimately degrade based upon both biotic and abiotic degradation mechanisms and do not pose any bioaccumulation hazard.

² EPIWIN v3.10; calculated using MPBPWIN v1.40 (determined at 760 millimeter mercury [mmHg]).

³ Lide, 2001-2002.

⁴ Bingham, 2001.

⁵ O'Neil, 2001.

⁶ Lewis, 2000.

⁷ EPIWIN v3.10; calculated using KOWWIN v1.66.

⁸ Sangster, 1989.

⁹ EPIWIN v3.10; calculated using WSKOW v1.40.

Table 3. Results for Environmental Fate and Pathways

Environmental Fate and Pathways				
Test	M/C	BZM	M/C	PHM
Photodegradation & Atmospheric Oxidation:				
OH Rate Constant	C ¹	44.63 x 10 ⁻¹² cm ³ /molecule-sec	C ¹	11.32 x 10 ⁻¹² cm ³ /molecule-sec
OH Half Life	C ¹	2.876 Hrs	C ¹	11.34 Hrs
Stability in Water (Hydrolysis)		No Data Available		No Data Available
Transport/ Distribution • Fugacity • Estimated Koc: • Estimated BCF:	C ² C ³	See model results below (Table 3a) 518 16.32	C ² C ³	See model results below (Table 3b) 268 17.39
Biodegradation	M ⁴	40.7% in 28 days (Closed Bottle - OECD 301d)		No Data Available

¹EPIWIN v3.10; calculated using AOP Program v1.40.

A. Photodegradation – Atmospheric Oxidation

Values for BZM photodegradation and atmospheric oxidation were calculated based upon chemical structures using EPIWIN and are shown in Table 3. A calculated half-life for BZM of 2.876 hours and rate constant of 44.63 x 10⁻¹² cubic centimeter (cm³)/molecule-sec has been estimated for reaction with hydroxyradicals, compared to a calculated half-life for PHM of 11.34 hours and a rate constant of 11.32 x 10⁻¹² cm³/molecule-sec.

Summary: These results are sufficient for USEPA HPV Challenge Program, and no further testing is warranted.

B. Hydrolysis

BZM and PHM are both water soluble and are expected to be stable in water at environmentally relevant pHs. EPIWIN was unable to calculate a hydrolysis rate constant for either structure due to the absence of functional groups that are labile to hydrolysis.

²EPIWIN v3.10; calculated using PCKOC Program v1.66.

³EPIWIN v3.10; calculated using BCF Program v2.14.

⁴ Elf Atochem S.A. Benzyl mercaptan, determination de la biodegradabilite facile, essai en fioles fermees.centre d'application de levallois, le 11/09/96

Summary: Testing for Stability in Water (OECD Test Guideline 111) is recommended for BZM to fulfill this endpoint.

C. Chemical Transport and Distribution in the Environment (Fugacity Modeling)

Tables 3a and 3b summarize the Level III Fugacity results for BZM and PHM produced by EPIWIN.

Table 3a. EPIWIN Level III Fugacity Results for BZM

Compartment	100% to air	100% to water	100% to soil	Equally to each compartment
Air	92.6%	0.87%	0.22%	1.54%
Water	6.13%	98.6%	1.85%	36.0%
Soil	1.2%	0.01%	97.9%	62.3%
Sediment	0.03%	0.51%	0.01%	0.19%

Table 3b. EPIWIN Level III Fugacity Results for PHM

Compartment	100% to air	100% to water	100% to soil	Equally to each compartment
Air	94.4%	3.43%	1.01%	5.37%
Water	4.58%	96.0%	1.61%	34.2%
Soil	0.99%	0.04%	97.4%	60.3%
Sediment	0.03%	0.52%	0.01%	0.19%

Summary: Results from the Level III fugacity modeling indicate that releases to water would remain in water while releases to air and soil would partition to water and soil. These results also show that both compounds behave similarly in the environment and that further fugacity modeling is not warranted.

D. Biodegradation and Bioaccumulation

BZM has been tested in a Ready Biodegradation test, and the results are reliable without restrictions and fulfill the HPV SIDS endpoint for BZM. The results are also in agreement with EPIWIN calculated results. BZM should be inherently biodegradable under real-world aerobic and anaerobic conditions. However, under the conservative conditions of the standard OECD ready tests, BZM was shown not to be readily biodegradable.

The EPIWIN predicted bioconcentration factor (BCF) and organic carbon partition coefficients (Koc) are similar for BZM and PHM, suggesting that both are nonsorptive in the environment and pose no bioaccumulation hazard.

Summary: These results are sufficient for USEPA HPV Challenge Program purposes, and no further testing is warranted (See Table 3 and IUCLID Documents).

Environmental Fate and Pathways Summary: Adequate data (i.e., Klimisch rating 1 and 2) are available for all environmental fate endpoints with the exception of Stability in Water (Hydrolysis), for which testing is proposed consistent with OECD Test Guideline 111.

VI. ECOTOXICITY DATA

Table 4 shows that only limited experimental aquatic toxicity testing of BZM and PHM has been conducted. One aquatic study was performed on aquatic invertebrates (*Daphnia*). However, within EPIWIN, the ECOSAR module recognizes BZM as a member of both the Neutral Organic and Thiols (or Mercaptan) chemical classes and recognizes PHM as a member of both the Neutral Organic and Phenols chemical classes. ECOSAR has validated fish Quantitative Structure Activity Relationship (QSAR) and *Daphnia* acute toxicity endpoints for these classes that add perspective, but clear trends are not evident.

Table 4. Results for Ecotoxicity Endpoints

Ecotoxicity				
Test	BZM	РНМ		
AcuteToxicity to Fish	96-hr LC_{50} = 64 mg/L ¹ 96-hr LC_{50} = 0.920 mg/L ²	96-hr $LC_{50} = 37 \text{ mg/L}^1$ 96-hr $LC_{50} = 6.082 \text{ mg/L}^3$		
Acute Toxicity to Aquatic Invertebrates (Daphnid)	24 hr EC ₅₀ > 0.26 mg/L ⁴ 48 hr EC ₅₀ = 0.15 mg/L ⁴ 48-hr EC ₅₀ = 0.045 mg/L ²	48-hr EC ₅₀ = 3.097 mg/L^3		
Acute Toxicity to Aquatic Plants (Algae)	No Data Available	96-hr EC ₅₀ = 13.4 mg/L ³		

¹ EPIWIN v3.10; calculated using ECOSAR Program for neutral organic chemicals.

Ecotoxicity Summary: Sufficient data is available to warrant no further acute toxicity testing with aquatic invertebrates. Additional testing is proposed (OECD Guidelines 203 and 201) to meet the Acute Fish and Acute Aquatic Plants (Algae) HPV SIDS endpoints.

² EPIWIN v3.10; calculated using ECOSAR Program for mercaptans.

³ EPIWIN v3.10; calculated using ECOSAR Program for phenols.

⁴ Elf Atochem S.A. 1997.

VII. MAMMALIAN TOXICITY

A limited amount of existing mammalian toxicity information on BZM and PHM demonstrates that PHM is a conservative read across benchmark, with slightly higher toxicity than BZM. The increased toxicity of PHM is likely due to its physicochemical characteristics; PHM has a slightly lower molecular weight, and is slightly more water soluble, mobile, and bioavailable than BZM. This is important because the trend is consistent and shows that where there are BZM data gaps in mammalian toxicity information, and PHM data are available for read-across purposes, additional testing on BZM will not provide new useful data.

Table 5. Results for Mammalian Toxicity Endpoints

Mammalian Toxicity				
Test	BZM	РНМ		
Acute Oral	LD ₅₀ 1 day = 985 mg/kg ¹ LD ₅₀ 15 day = 493 mg/kg ¹	$LD_{50} = 46.2 \text{ mg/kg}^1$		
Acute Inhalation	A. Rat LC ₅₀ : not calculable (one death out of 6 rats at 185 ppm and one death out of 6 rats at 235 ppm. ¹	A. Rat LC ₅₀ (48 hr): 59 ppm ¹ Rat LC ₅₀ (15 day): 33 ppm ¹		
IIIIIaiauoii	B. Mouse LC ₅₀ (24 hr): 195 ppm ¹ Mouse LC ₅₀ (15 day): 178 ppm ¹	B. Mouse LC ₅₀ (24 hr): 47 ppm ¹ Mouse LC ₅₀ (48 hr): 35.5 ppm ¹ Mouse LC ₅₀ (15 day): 28 ppm ¹		
Acute Dermal	Rat LD_0 : $\geq 2000 \text{ mg/kg}^2$	Rat LD ₅₀ : 300 mg/kg ¹		
Acute (i.p.)	LD ₅₀ 1 day = 429 mg/kg ¹ LD ₅₀ 15 day = 373 mg/kg ¹	LD ₅₀ 1 day = 25.2 mg/kg ¹ LD ₅₀ 15 day = 9.8 mg/kg ¹		
Repeated Dose	No Data Available	No Data Available		
Reproduction Toxicity	No Data Available	Reproductive toxicant both in males (based on increased incidence of inhibited spermiation in all treated F ₁ males, and decreased epididymal sperm motility in the mid- and high dose [18 and 35 mg/kg] F ₀ males) and in females (developmental based on decreased pup weights). ³		
Developmental Toxicity	No Data Available	A. Rat Maternal LOAEL: 20 mg/kg/day ⁴ ; Rat Fetal NOAEL: 20 mg/kg/day ⁴ B. Rabbit Maternal NOAEL: 10 mg/kg/day ⁵ ; Rabbit Fetal NOAEL: 40 mg/kg/day ⁵		
Genetic – Gene Mutation	Negative ⁶	No Data Available		

Mammalian Toxicity				
Test	BZM	PHM		
Genetic – Chromosomal Aberration	No Data Available	No Data Available		

¹ Fairchild and Stokinger, 1958.

LOAEL = lowest observed adverse effect level.

NOAEL = no observed adverse effect level.

A. Acute Toxicity

Acute toxicity studies show that BZM is of low acute toxicity by the oral, inhalation, dermal, and interperitoneal routes (see Table 5 and IUCLID documents). Importantly, the close structural analogue PHM was also tested and consistently shows higher acute toxicity than BZM.

Summary: These studies fulfill the HPV requirements for the acute toxicity endpoint; no additional testing is proposed for the USEPA HPV Challenge Program.

B. Repeated Dose Toxicity

No repeated dose toxicity studies were identified for either BZM or its structural surrogate, PHM. As discussed in other sections, BZM will not be considered a closed system intermediate and therefore testing for this endpoint is proposed.

Summary: Additional testing is proposed for this endpoint consistent with OECD Guideline 407 (Repeated Dose 28-day Oral Toxicity Study in Rodents).

C. Genetic Toxicity/Mutagenicity

A valid *in vitro* gene mutation study (*Salmonella typhimurium* Reverse Mutation Assay) was performed for BZM and showed no mutagenic activity. No *in vitro* chromosomal aberration studies were identified for either BZM or PHM. The USEPA requires two different endpoints to be tested: gene mutation and chromosomal aberration. As a result, testing is proposed to meet the chromosomal aberration requirements.

Summary: Additional testing is proposed for this endpoint consistent with OECD Guideline 473 (Genetic Toxicology: *In vitro* Mammalian Cytogenetic Test) in order to meet the chromosomal aberration endpoint requirements.

² Centre International de Toxicologie, 1996.

³ National Toxicology Program, 1996.

⁴ National Toxicology Program, 1994a.

⁵ National Toxicology Program, 1994b.

⁶ Wild et al., 1983.

D. Reproductive Toxicity

BZM has not been tested for reproductive toxicity. However, due to its close structural similarity to PHM, as well as the demonstrated higher level of acute toxicity of PHM, it would be expected that BZM would be of a similar order of magnitude as PHM, if not of lower reproductive toxicity. As a result, data for the structural surrogate PHM can be used as read-across data for BZM, and no further testing of BZM is warranted for this endpoint (see Table 5 and IUCLID Documents).

PHM was tested by the National Toxicology Program (1996) for reproductive toxicity in rats. This study followed the National Toxicology Program's Reproductive Assessment by Continuous Breeding (RACB) protocol and was found to be valid without restriction. PHM was determined to be a slight reproductive toxicant both in males (based on increased incidence of inhibited spermiation in all treated F_1 males, and decreased epididymal sperm motility in the mid- and high dose [18 and 35 mg/kg] F_0 males) and in females (developmental – based on decreased pup weights). Results of this study also show that PHM is not a selective reproductive toxicant because the minor effects on reproduction occurred concomitant with, or at doses greater than, those doses that produce hepatic or renal toxicity.

Summary: Adequate data (i.e., Klimisch rating 1) are available for this endpoint. Data for the structural surrogate PHM can be used as read-across data for BZM, and no further testing of BZM is warranted for this endpoint (see Table 5 and IUCLID Documents).

E. Developmental Toxicity

PHM was tested by the National Toxicology Program (1994a; 1994b) for developmental toxicity in both rats and rabbits. These studies followed OECD Guideline 414 and were found to be valid without restriction (Klimisch 1). In rats, the maternal LOAEL was 20 mg/kg/day, and the fetal NOAEL was 20 mg/kg/day. Maternal toxicity (observed as maternal mortality), a persistent decrease in body weight and weight gain, and a decrease in food consumption during the treatment period occurred at the high-dose level of 50 mg/kg/day. The LOAEL (20 mg/kg/day) for maternal toxicity was based on minor, transient decreases in maternal weight gain and food consumption on gestation day (gd) 6 to 9. The maternal NOAEL could not be determined based upon the doses evaluated. Developmental toxicity, observed as increased post-implantation death, decreased litter size, decreased fetal body weight, and an increase in the incidence of external malformations, occurred only at the high dose. Reduced female fetal body weight was observed at 35 mg/kg/day, suggesting an NOAEL of 20 mg/kg/day.

In rabbits, the maternal NOAEL was 10 mg/kg/day, and the fetal NOAEL was 40 mg/kg/day. The 40 mg/kg/day PHM did not adversely affect the growth, viability, or morphological development of the offspring. As a result, the developmental toxicity in this study was \geq 40 mg/kg/day; the LOAEL could not be determined at the doses

evaluated. Maternal toxic effects at 30 and 40 mg/kg/day were minor and transient; therefore, the evidence of toxicity was equivocal. However, a slightly higher dose of 50 mg/kg/day was found to be excessively toxic, resulting in maternal morbidity and mortality. Evaluation of developmental toxicity at doses above 40 mg/kg/day was precluded by excessive maternal toxicity.

BZM has not been tested for developmental toxicity. However, due to its close structural similarity to PHM, as well as the demonstrated higher level of acute toxicity of PHM, it would be expected that BZM would be of a similar order of magnitude as PHM, if not of lower developmental toxicity.

Summary: Data for the structural surrogate PHM can be used as read-across data for BZM, and no further testing of BZM is warranted for this endpoint (see Table 5 and IUCLID Documents).

Mammalian Toxicity Summary: Sufficient mammalian toxicity data exist for BZM and PHM for all SIDS Mammalian Toxicity Endpoints except repeated dose and genetic toxicity. Additional testing (OECD Guidelines 407 and 473) is proposed to fulfill these two SIDS endpoints.

VIII. "BEYOND SIDS" ENDPOINTS:

Studies have been performed to investigate skin irritation and skin sensitization potential (see IUCLID document).

IX. CONCLUSIONS

As summarized below, CPChem concludes that there are sufficient, reliable data on BZM and its structural surrogate, PHM, for many of the SIDS endpoints following a thorough review of company proprietary files, the peer-reviewed literature, and/or calculations using widely accepted computer modeling programs.

- **PHYSICOCHEMICAL DATA**. Physicochemical endpoints for BZM are fulfilled by using existing measured data or data calculated by the EPIWIN[©] computer model. No additional testing is proposed.
- ENVIRONMENTAL FATE. Sufficient data are available to characterize environmental fate endpoints, with the exception of hydrolysis, for BZM. An estimation from a Level III fugacity model predicts that both BZM, as well as its structural surrogate PHM, will likely partition to soil and water. Ready biodegradation testing showed that BZM is not readily biodegradable and the predicted bioconcentration factors and organic carbon partition coefficients for BZM and PHM suggest similar fate profiles in the environment and a slight

bioaccumulation hazard for either BZM or PHM. Additional testing for the hydrolysis endpoint is proposed.

- ACUTE AQUATIC TOXICITY. Only limited acute aquatic toxicity data are available for BZM and PHM. Additional acute toxicity testing for fish and aquatic plants is proposed for BZM.
- ACUTE MAMMALIAN TOXICITY. Mammalian toxicity data demonstrates a low order of toxicity via oral, dermal, and inhalation routes of exposure. Sufficient data are available to fulfill the acute toxicity endpoint for BZM and no additional testing is proposed.
- **GENETIC TOXICITY**. Limited genetic toxicity data is available for BZM. Additional testing using an *in vitro* chromosomal aberration assay is proposed.
- **REPEATED DOSE TOXICITY**. No repeated dose toxicity testing data for BZM or PHM was identified. Additional testing for this endpoint is proposed.
- REPRODUCTIVE AND DEVELOPMENTAL TOXICITY. BZM has not been tested for reproductive and developmental toxicity, but, due to its close structural similarity to PHM, it would be expected to be of a similar order of magnitude as PHM. Given that PHM is characterized for this endpoint and is expected to produce results of the same order of magnitude, PHM can be used as a structural surrogate for BZM, and no further BZM testing is warranted for this endpoint.

Test Plan Summary: Additional testing with BZM is proposed to fulfill the following five endpoints:

- Hydrolysis (OECD 111);
- Acute Toxicity to Fish (OECD 203);
- Acute Toxicity to Aquatic Plants (OECD 201);
- Repeated Dose (OECD 407); and
- Genetic Toxicity: In vitro Mammalian Cytogenetic Test (OECD 473).

X. REFERENCES

Bingham, E. 2001. *Patty's Toxicology*, Fifth Edition, Vol. 7. New York: John Wiley & Sons, Inc.

Centre International de Toxicologie (C.I.T.). 1996. *Acute Dermal Toxicity in Rats – Benzyl Mercaptan*. Laboratory Study Number 12562-TAR. Miserey, France.

Elf Atochem SA. Benzyl mercaptan, determination de la biodegradabilite facile, essai en fioles fermees centre d'application de levallois, le 11/09/96

Elf Atochem SA. Benzylmercaptan, toxicite aiguë vis-a-vis des daphnies.centre d'application de levallois, etude n°: 4353/96/a, le 31/01/97.

Fairchild, E.J., and H.E. Stokinger. 1958. Toxicologic studies on organic sulfur compounds. 1. Acute toxicity of some Aliphatic and Aromatic Thiols (Mercaptans). *Industrial Hygiene Journal* (June).

Klimisch, H.J., E. Andreae, and U. Tillmann. 1997. A systematic approach for evaluating the quality of experimental and ecotoxicological data. *Reg. Tox. and Pharm*. 25: 1-5.

Lewis, R.J., Sr. 2000. Sax's Dangerous Properties of Industrial Materials, Tenth Edition. New York: John Wiley & Sons, Inc.

Lide, D.R. 2001-2002. *CRC Handbook of Chemistry and Physics*, 82nd Edition. Boca Raton, FL: CRC Press, Inc.

National Toxicology Program. 1994a. Final Report on the Developmental Toxicity of Thiophenol (CAS #108-98-5) in Sprague-Dawley CD Rats. U.S. Department of Health and Human Services, Public Health Services, National Institutes of Health. TER92133.

National Toxicology Program. 1994b. Final Report on the Developmental Toxicity of Thiophenol (CAS #108-98-5) in New Zealand White (NZW) Rabbits. U.S. Department of Health and Human Services, Public Health Services, National Institutes of Health. TER92134.

National Toxicology Program. 1996. Final Report on the Reproductive Toxicity of Thiophenol (CAS #108-98-5) Administered by Gavage to Sprague-Dawley Rats. U.S. Department of Health and Human Services, Public Health Services, National Institutes of Health. R.O.W. Sciences 8989-24.

Organisation for Economic Co-operation and Development (OECD) Secretariat. 2002. *Manual for Investigation of HPV Chemicals* (November 2002).

O'Neil, M.J. 2001. *The Merck Index: An Encyclopedia of Chemicals, Drugs and Biologicals*, 13th Edition. Whitehouse Station NJ: Merck & Co., Inc.

Sangster, J. 1989. Octanol-water partition coefficients of simple organic compounds. *J. Phys. Chem. Ref. Data* 18(3): 1111-11120.

United States Environmental Protection Agency (USEPA), Office of Pollution Prevention and Toxics. 1999a. Draft Determining the Adequacy of Existing Data. Washington, D.C.

USEPA. 1999b. The Use of Structure-Activity Relationships (SAR) in the High Production Volume Chemicals Challenge Program. USEPA, Office of Pollution Prevention and Toxics. Washington, D.C.

USEPA (Office of Pollution Prevention and Toxics) and Syracuse Research Corporation. 2000. EPI Suite v 3.10 (April, 2001).

Wild, D., M.T. King, E. Gocke, and K. Eckhardt. 1983. Study of artificial flavouring substances for mutagenicity in the Salmonella/Microsome, BASC and Micronucleus Tests. *Fd. Chem. Toxic.* 21(6): 707-719.